

"Grade or Education" = 1

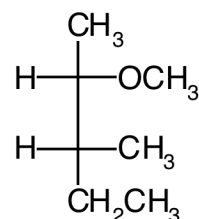
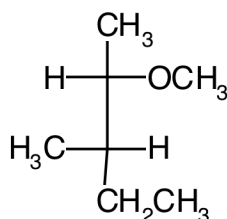
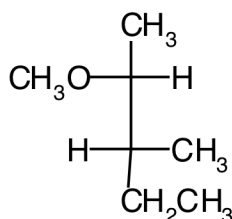
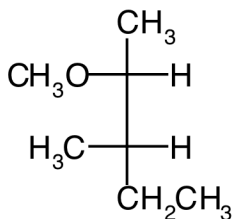
CHEM 2261/01

Summer 08

Exam 3

Chapters 7-9

1. Find the substitution product(s) of the reaction of (2S,3R)-2-chloro-3-methylpentane + high concentration of CH_3O^- from among the numbered structures shown below. Choose the CORRECT statement from the multiple choices.

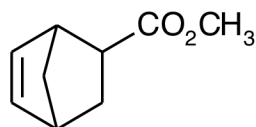


- ☐ A. 1 is the sole product of this reaction.
- ☐ B. 3 is the sole product of this reaction.
- ☐ C. 2 is the sole product of this reaction.
- ☐ D. 2 and 4 are both products of this reaction.
- ☐ E. 4 is the sole product of this reaction.

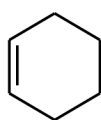
Rationale:

Chapter 8 Problem 52b

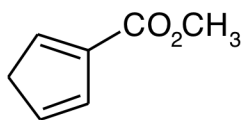
2. What two compounds could be used to produce the compound whose structure is shown below from a Diels-Alder reaction?



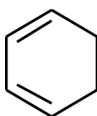
Pick the structures of the two compounds which can be used in this Diels-Alder reaction from the numbered choices below.



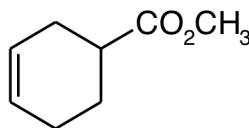
1



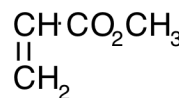
2



3



4



5



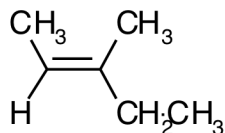
6

- ___ A. Compounds 5 and 6 can be used in this reaction.
 ___ B. Compounds 1 and 2 can be used in this reaction.
 ___ C. Compounds 4 and 6 can be used in this reaction.
 ___ D. Compounds 3 and 5 can be used in this reaction.
 ___ E. Compounds 3 and 4 can be used in this reaction.

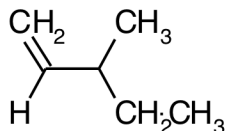
Rationale:

Chapter 7 Problem 64b

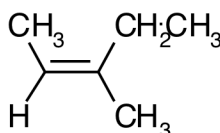
3. Find the elimination product(s) of (2S,3R)-2-chloro-3-methylpentane + high concentration of CH_3O^- among the numbered structures below. Choose the CORRECT product of this reaction.



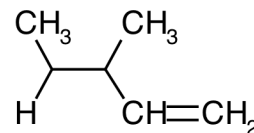
1



2



3



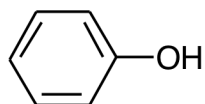
4

- ___ A. 1 and 3 are both products of this reaction.
 ___ B. 4 is the product of this reaction.
 ___ C. 2 is the product of this reaction.
 ___ D. 1 is the product of this reaction.
 ___ E. 3 is the product of this reaction.

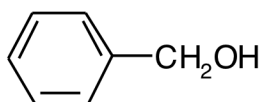
Rationale:

Chapter 9 Problem 48b

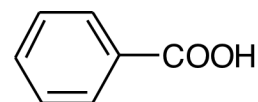
4. Rank the compounds whose structures are shown below in order of INCREASING acid strength (weakest acid listed first).



1



2



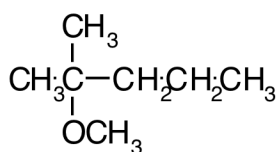
3

- ___ A. 1 < 3 < 2
 ___ B. 1 < 2 < 3
 ___ C. 3 < 2 < 1
 ___ D. 3 < 1 < 2
 ___ E. 2 < 1 < 3

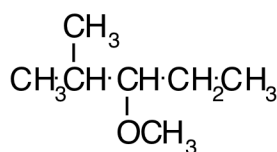
Rationale:

Chapter 7 Problem 18

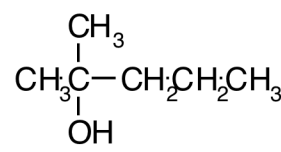
5. Choose the CORRECT substitution product of the reaction of 3-bromo-2-methylpentane + CH_3OH from the numbered structures shown below.



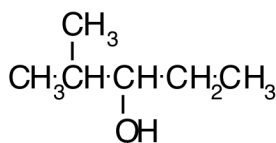
1



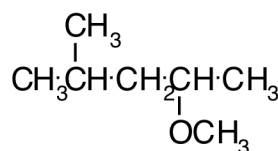
2



3



4



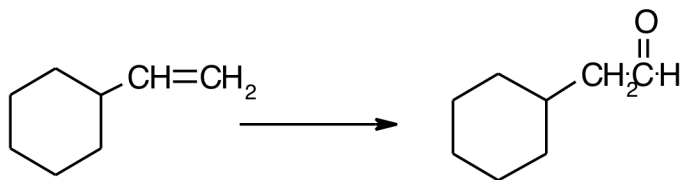
5

- ___ A. 4 is the product of this reaction.
 ___ B. 5 is the product of this reaction.
 ___ C. 3 is the product of this reaction.
 ___ D. 1 is the product of this reaction.
 ___ E. 2 is the product of this reaction.

Rationale:

Chapter 8 Problem 45e

6. For the target compound whose structure is shown below choose the multistep synthesis which could be used to prepare it from the given starting material.

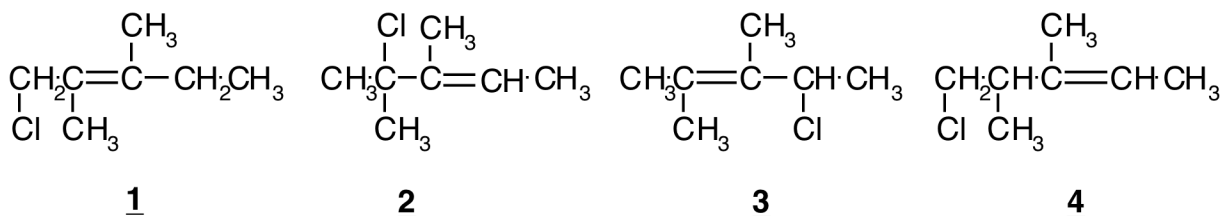


- ___ A. First: $\text{Br}_2/\text{CH}_2\text{Cl}_2$; Next: excess NH_2 ; Finally: $\text{H}_2\text{SO}_4/\text{HgSO}_4$
 ___ B. First: $\text{Br}_2/\text{CH}_2\text{Cl}_2$; Next: excess NH_2 ; Finally: 1. disiamylborane and 2. HO^- , H_2O_2 , H_2O
 ___ C. First: $\text{HBr}/\text{CH}_2\text{Cl}_2$; Next: NH_2 ; Finally: 1. BH_3/THF and 2. HO^- , H_2O_2 , H_2O
 ___ D. First: $\text{Br}_2/\text{CH}_2\text{Cl}_2$; Next: excess NH_2 ; Finally: $\text{H}_2\text{SO}_4/\text{H}_2\text{O}$
 ___ E. First: $\text{Br}_2/\text{H}_2\text{O}$; Next: NH_2

Rationale:

Chapter 9 Problem 31b

7. Find the kinetic and thermodynamic products of the reaction of one equivalent of HCl with 2,3-dimethyl-1,3-pentadiene among the numbered structures below. Choose the CORRECT statement from the multiple choices.

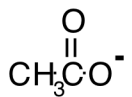


- ___ A. 4 is the kinetic product and 1 is the thermodynamic product.
 ___ B. 1 is the kinetic product and 4 is the thermodynamic product.
 ___ C. 3 is the kinetic product and 1 is the thermodynamic product.
 ___ D. 2 is the kinetic product and 3 is the thermodynamic product.
 ___ E. 3 is the kinetic product and 2 is the thermodynamic product.

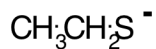
Rationale:

Chapter 7 Problem 68a

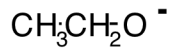
8. Rank the ions whose structures are shown below in order of DECREASING nucleophilicity in methanol (strongest nucleophile listed first).



1



2



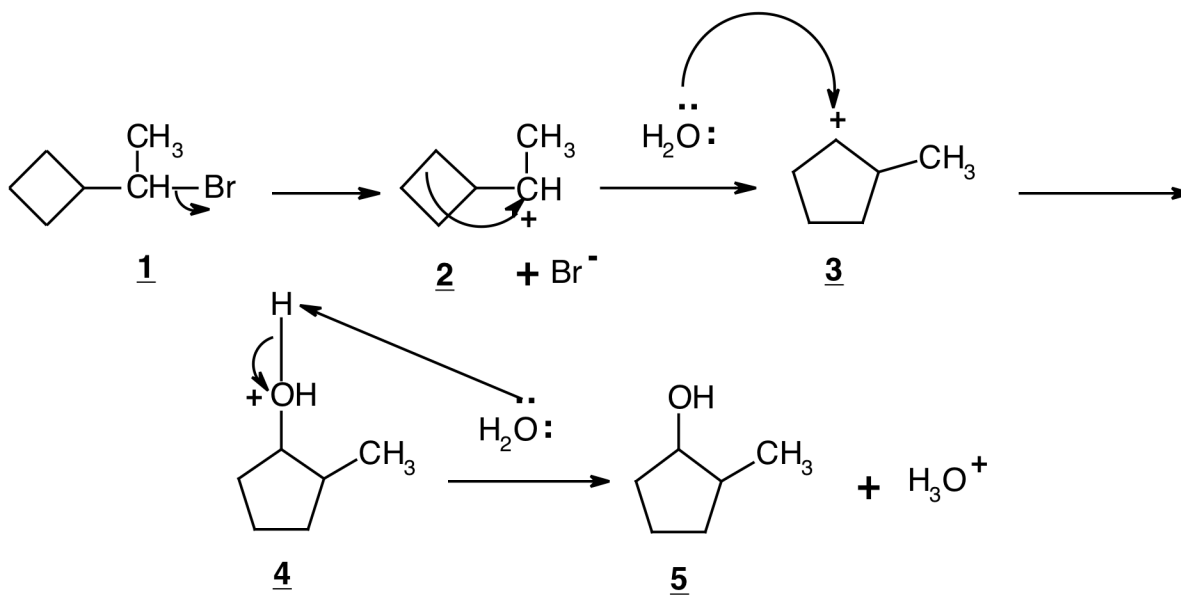
3

- ___ A. 2 > 3 > 1
 ___ B. 1 > 3 > 2
 ___ C. 1 > 2 > 3
 ___ D. 3 > 2 > 1
 ___ E. 2 > 1 > 3

Rationale:

Chaptr 8 Problem 42a

9. Shown below is a curved-arrow mechanism for converting structure 1 into structure 5. Which structure has the curved arrow(s) associated with it drawn INCORRECTLY?

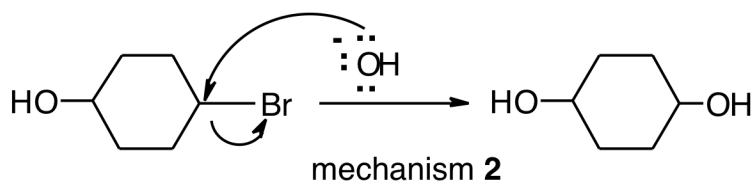
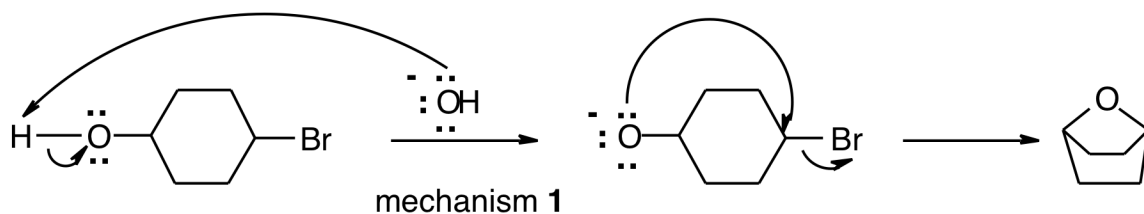


- ___ A. 3
 ___ B. 5
 ___ C. 4
 ___ D. 1
 ___ E. 2

Rationale:

Chapter 8 Problem 62a

10. Two curved-arrow mechanisms are shown for the reaction of 4-bromocyclohexanol with HO^- to form substitution products. Figure out the substitution mechanism(s) for the reaction of cis-4-bromocyclohexanol and trans-4-bromocyclohexanol with HO^- to form product(s). Choose the CORRECT statement from the multiple choices.

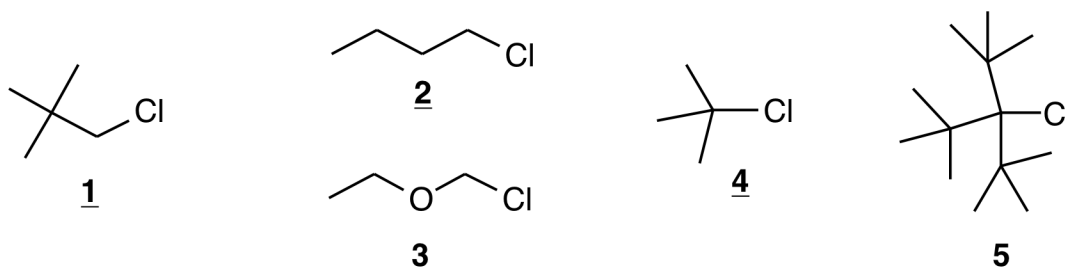


- ___ A. cis-4-bromocyclohexanol undergoes mechanism 2.
- ___ B. Neither cis-4-bromocyclohexanol nor trans-4-bromocyclohexanol undergo mechanism 2.
- ___ C. cis-4-bromocyclohexanol undergoes mechanism 1.
- ___ D. trans-4-bromocyclohexanol cannot react by either mechanism.
- ___ E. Both cis-4-bromocyclohexanol and trans-4-bromocyclohexanol undergo mechanism 1.

Rationale:

Chapter 9 Problem 55a

11. Use the numbered structures shown below to choose the CORRECT statement from the multiple choices.

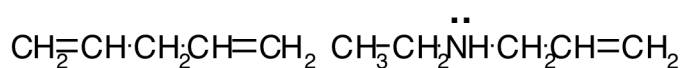


- ___ A. $(\text{CH}_3)_3\text{CBr}$ will react faster with $\text{CH}_3\text{CH}_2\text{OH}$ than it will with H_2O .
- ___ B. $(\text{CH}_3)_2\text{CHS}^-$ will react faster with 1 than CH_3S^- will.
- ___ C. 3 will not react with HO^- .
- ___ D. 5 will react with H_2O faster than 4 will.
- ___ E. 2 will react with HO^- faster than 3 will.

Rationale:

Chapter 8 Problem 48

12. Which of the structures shown below has delocalized electrons?

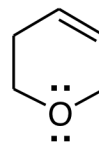


1

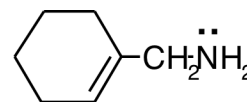
2



3



4



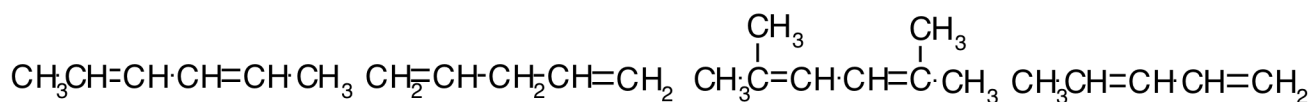
5

- ___ A. structure 4
 ___ B. structure 3
 ___ C. structure 5
 ___ D. structure 2
 ___ E. structure 1

Rationale:

Chapter 7 Problem 3

13. Name the dienes shown below and rank them in order of INCREASING stability (name of least stable diene listed first).

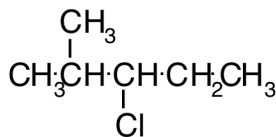


- ___ A. 1,5-pentadiene < 2,5-pentadiene < 2,5-hexadiene < 2,5-dimethyl-2,5-hexadiene
 ___ B. 1,4-pentadiene < 1,3-pentadiene < 2,4-hexadiene < 2,5-dimethyl-2,4-hexadiene
 ___ C. 1,3-pentadiene < 2,4-hexadiene < 1,4-pentadiene < 2,5-dimethyl-2,4-hexadiene
 ___ D. 2,5-dimethyl-2,4-hexadiene < 2,4-hexadiene < 1,3-pentadiene < 1,4-pentadiene
 ___ E. 2,5-dimethyl-2,5-hexadiene < 2,5-hexadiene < 2,5-pentadiene < 1,5-pentadiene

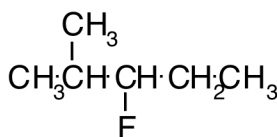
Rationale:

Chapter 7 Problem 10

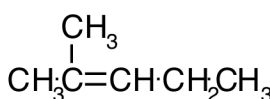
14. Figure out what the major elimination product(s) obtained from E2 reactions of hydroxide ion and the alkyl halides with structures 1 and 2 are. Choose the statement which is CORRECT from the multiple choices.



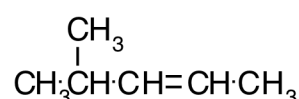
1



2



3



4

- ___ A. 1 reacts with hydroxide to give 3 as the major E2 product and 2 reacts with hydroxide to give 4 as the major E2 product.
- ___ B. Both 1 and 2 react with hydroxide to give 3 as the major E2 product.
- ___ C. Both 1 and 2 react with hydroxide to give 4 as the major E2 product.
- ___ D. 1 reacts with hydroxide to give 4 as the major E2 product and 2 reacts with hydroxide to give 3 as the major E2 product.
- ___ E. 1 and 2 react with hydroxide to give E2 products which are not shown in the structures above.

Rationale:

Chapter 9 Problem 4(b,f)

15. By looking at the numbered structures below find the major elimination product(s) of a mixture of the two stereoisomers of trans-1-chloro-2-methylcyclohexane for the conditions specified in each of the multiple choices. Choose the CORRECT statement.



1



2



3

- ___ A. trans-1-chloro-2-methylcyclohexane + high concentration of CH_3O^- yields 2.
- ___ B. trans-1-chloro-2-methylcyclohexane + high concentration of CH_3O^- yields 1 and 2.
- ___ C. trans-1-chloro-2-methylcyclohexane + CH_3OH yields 2 and 3.
- ___ D. trans-1-chloro-2-methylcyclohexane + high concentration of CH_3O^- yields 1 and 3.
- ___ E. trans-1-chloro-2-methylcyclohexane + CH_3OH yields 1 and 3.

Rationale:

Chapter 9 Problem 38(c,d)

Answer Key

"Grade or Education" = 1

**CHEM 2261/01
Summer 08
Exam 3
Chapters 7-9**

1. C
2. A
3. E
4. E
5. D
6. B
7. D
8. A
9. E
10. A
11. D
12. B
13. B
14. A
15. D